Evaluation of Alternatives to Domestic Ion Exchange Water Softeners

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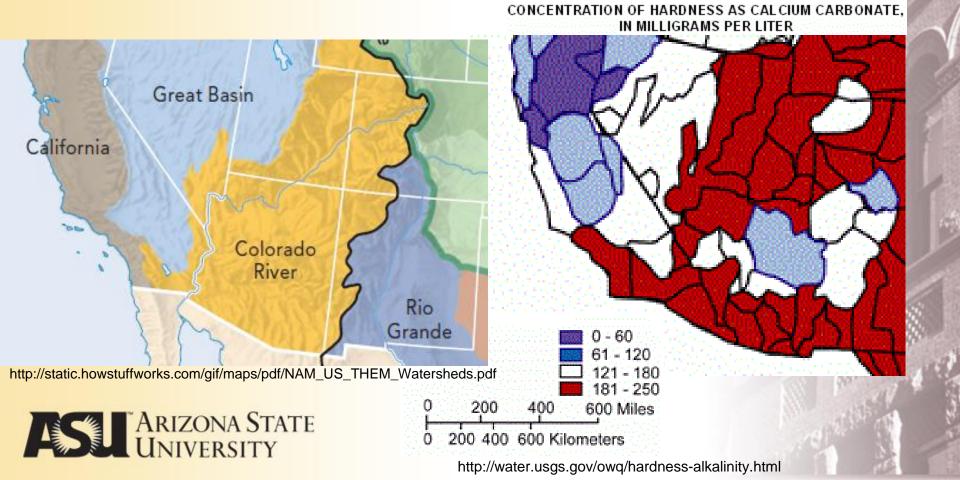
April 26, 2011

OUTLINE

- Water Quality and reuse in the Southwest US
- Ion exchange water softening system process and effects on remediated water quality
- No-salt alternatives to ion exchange and the mechanisms by which they reduce scale formation
- Experimental procedure
- Results
- Future Work



 Freshwater sources in the Southwest US are considered very hard ranging from 80 to 280 mg/L.



Hard Water Effects in the Home



•Spotted dishes from the dishwasher •An inability for soap to lather and soap scum deposits •Scale formation on faucets and showerheads Scale accumulation in pipes •Scale fouling in water heaters increasing energy usage by up to 24% •Scale formation on appliances

Calcium carbonate becomes less soluble at higher temperatures.



- Consumers try to mitigate the effects of hard water by using water softening devices in their homes.
- The most common domestic water softening device uses ion exchange technology which releases additional salts to the waste stream.
- Consumers are reducing hardness in their homes but increasing TDS levels in reclaimed wastewater! (Not a sustainable practice)



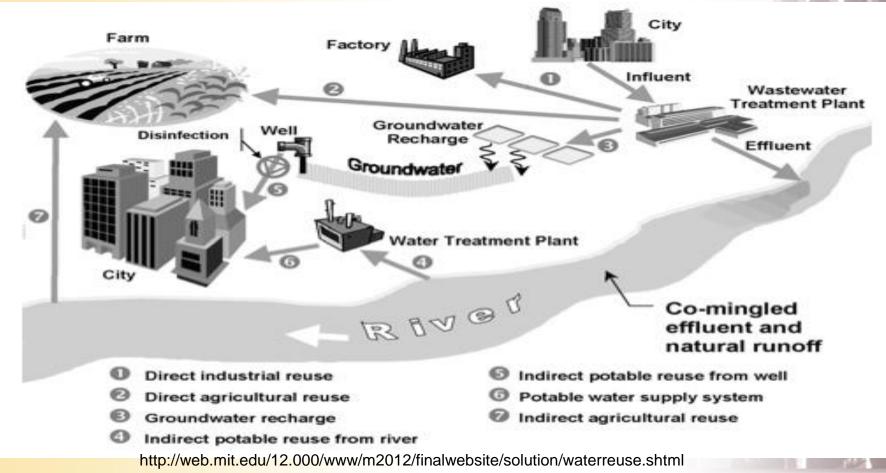
 TDS (salinity) is a measurement of total dissolved solids in water including inorganic (hardness, salts) and organic substances (pesticides, herbicides, etc.).

Sources of Salinity

- Natural minerals in rocks found in lakes, rivers, streams and aquifers
- Water from natural salt springs that enters into rivers, lakes and streams
- Agricultural fertilizers that drain from fields into rivers, lakes, streams and aquifers
- Water treatment chemicals such as chlorine that make water safe for human consumption
- Home water treatment systems, like water softeners, that treat water for hardness
- Cleaning chemicals
- Foods

Water Source	TDS in milligrams per liter
Salt River	580 mg/L
Verde river	270 mg/L
Central Arizona Project (CAP)	650 mg/L
Groundwater	200 - 5,000 mg/L
Reclaimed Water	Typically 300 - 500 mg/L higher than source water

 Water Reuse
A water conservation practice in which reclaimed water is used for a direct beneficial purpose.



TDS effects on water reuse (examples)

- Agriculture
 - Crop salt tolerance, reduction of crop yields
 - Additional water may be needed to flush salts from root zone
- Cooling Tower
 - Increased water usage
 - Possible equipment damage due to scaling



Study Objective

Provide technical data to identify credible alternatives to ion exchange water softeners that would provide consumers with the ability to reduce the impacts of hard water without creating the negative salinity impacts.



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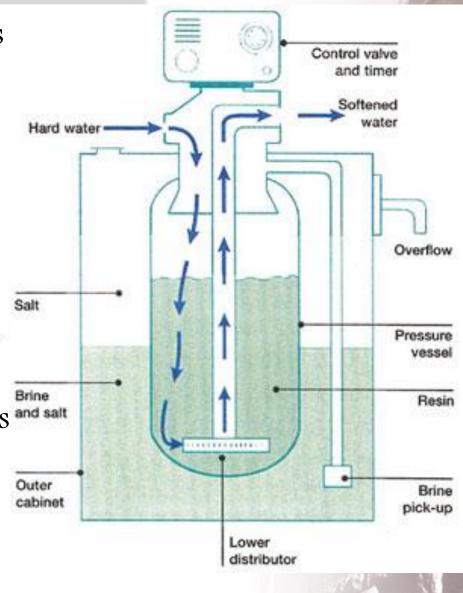


Ion Exchange Water Softening System

• The ion exchange unit removes hardness by exchanging sodium ions for the calcium and magnesium ions present in the water.

- It does this using resin beads that periodically need to be regenerated with a highly concentrated salt solution.
- There are two basic types of self-regenerating water softeners (SRWS): Timer Based and Demand Based.





Ion Exchange Water Softener Systems

- Discharge brine into wastewater systems
- These unnatural quantity of salts find their way into the environment and affect reuse applications.
- The use of no-salt water conditioning devices to reduce scale formation on domestic water heaters and other home appliances is one way society can improve the quality of remediated water.



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Capacitive Deionization

Electrically Induced Precipitation

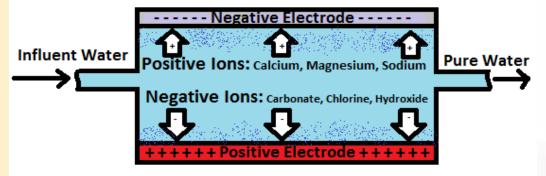
Template Assisted Crystallization

Electromagnetic Water Treatment

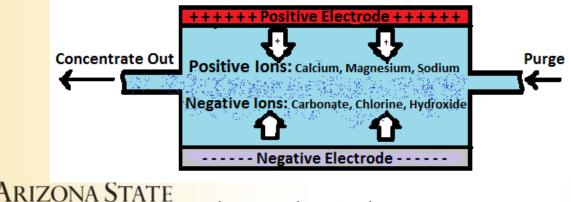


Capacitive Deionization

Regeneration: Voltage potential turned on

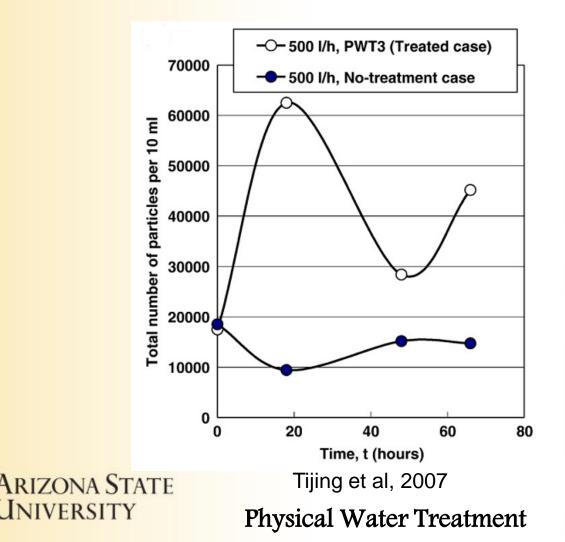


Backwash: Voltage potential turned off or reversed



Electro-chemical Treatment

Electronically Induced Precipitation



Template Assisted Crystallization

Mg and Ca ions collect on nucleation sites and form solid crystals that Mg won't attach to surfaces

Nucleation Sites

Polymer Bead



Electromagnetic Water Treatment





Physical Water Treatment

Other possible mechanisms for magnetic treatment

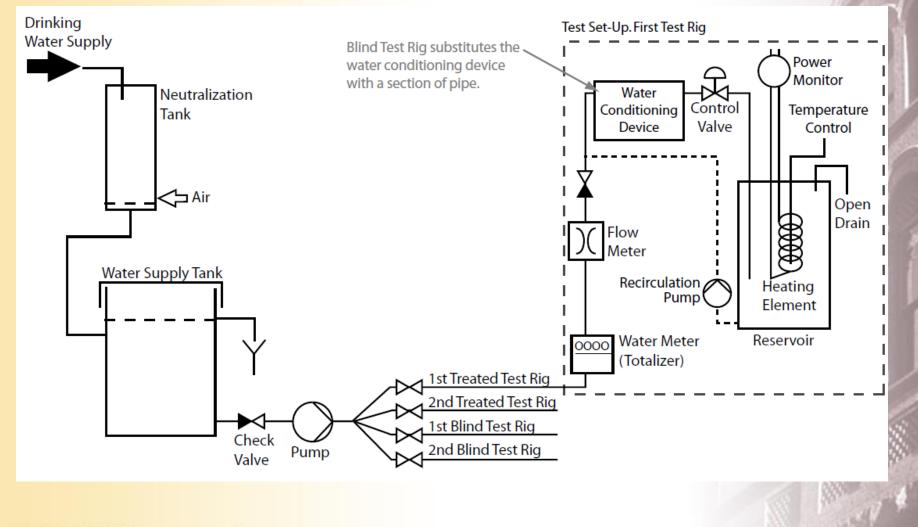
- Reduction of the effect of the double layer
 - When the electrical double layer is reduced, more suspended coagulation can occur resulting in a light sludge that is easily wiped off of the surface.
 - This can be tested by measuring the zeta potential of a particle before and after treatment.



OUTLINE

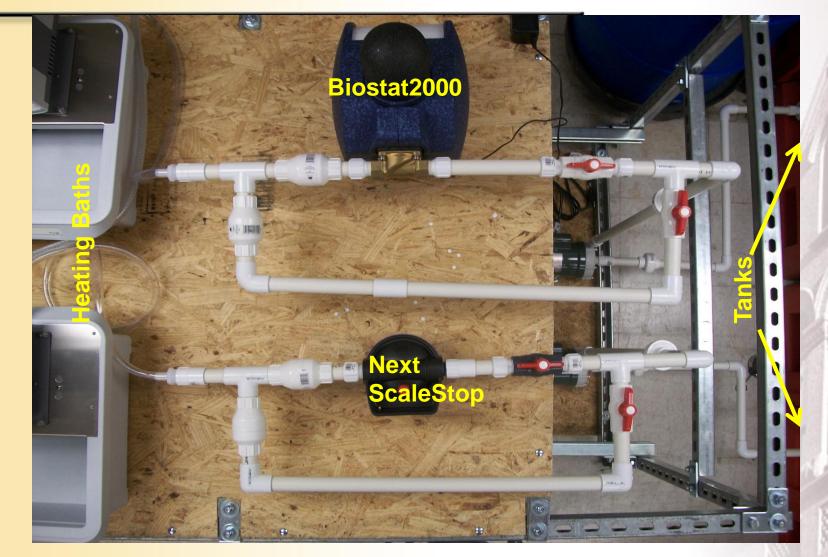
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DVGW-W512 Protocol





Plan View





- Once the 21 days of testing is over, the bath and heating element are cleaned using a 1N HCl solution.
- The solid scale is weighed and the scale dissolved by the HCI solution is measured using a Hach kit which utilizes the EDTA complexing method.
- This procedure will be repeated for all alternative devices using 3 different water qualities.
 CARIZONA STATE

Water Qualities Included

- Salt River water (Tempe tap water)
- Central Arizona Project (CAP) canal water
- Scottsdale groundwater

	TDS (mg/L)	Hardness (mg/L as CaCO ₃)
Salt River water (Tempe tap water)	479	180
Central Arizona Project (CAP) canal water	. 666	150 - 220
Scottsdale groundwater	465	200 - 250



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	Water Type	Treatment device used	Scale scraped off of heating element (g)	scale	(g Ca)	Scale from bath and heating element dissolved with HCl (g Ca as CaCO ₃)	test (g Ca as CaCO ₃)	Photo of heating element with scale
		No Treatment	-	NA	0.00	8.36	8.36	
	Tempe tap water	TAC	0.00	NA	0.00	0.12	0.12	
		EIP	0.68	34.88	0.24	3.60	3.84	
		MAG	1.44	34.88	0.50	3.47	3.97	
		CDI	0.00	NA	0.00	1.41	1.41	



ELECTROMAGNETIC

ELECTRICALLY INDUCED

CDI

RELEX

TAC

Mass Balance

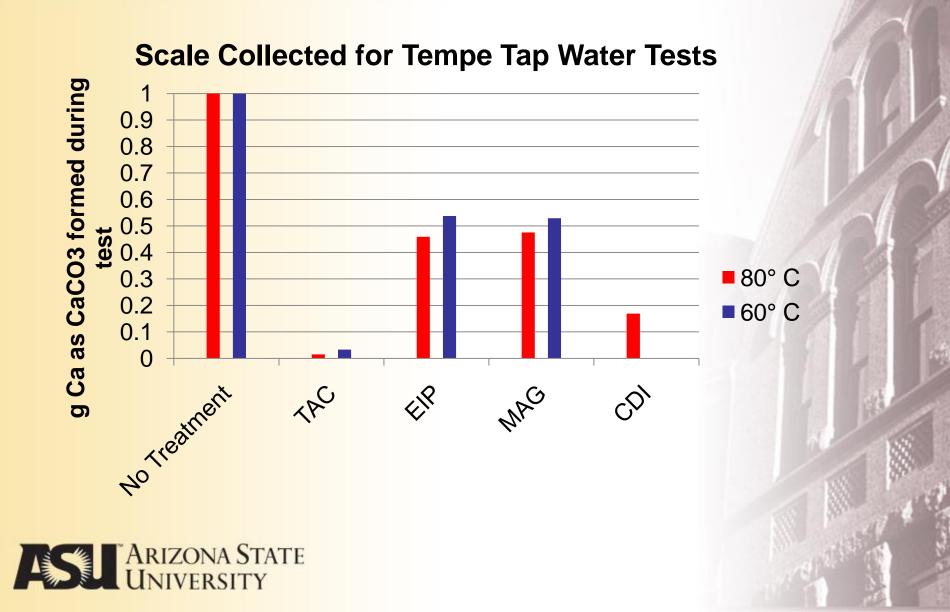
Treatment Device	Total Initial Ca as CaCO ₃ (g) Before Treatment*	Ca as CaCO ₃ Found on Heating Element and Bath (g)	Total effluent Ca as $CaCO_3$ exiting the system (g)	% Scale Formed on Heating Element and Bath
No Treatment	<mark>29</mark> 4	8.36	285.64	2.84%
TAC	<mark>29</mark> 4	0.31	293.69	0.11%
EIP	294	4.07	289.93	1.38%
MWT	<mark>29</mark> 4	4.86	289.14	1.65%
CDI	294	1.41	292.59	0.48%

*Initial Ca indicates the average calcium content in 700gal Tempe tap water



	Solid calcium Water Type Treatment collected device used from element (g Ca)		element dissolved	Scale from bath dissolved with 1N HCl (g Ca as CaCO3)	Total calcium formed during test (g Ca as CaCO₃)	d Photo of heating element after 21 days of testing	
=	Trea	No Treatment		5.92	19.00	24.92	
		TAC		0.83		0.83	
	water 60°C	EIP	0.33	5.88	7.19	13.40	
		MAG		6.20	7.00	13.20	

120



Percent Removal Compared to Untreated Case					
	Tempe Tap				
	80°C	60°C			
No Treatment	0	0			
TAC	99	97			
EIP	54	46			
MAG	53	47			
CDI	83	The second se			

To "pass" the DVGW-W512 test, a percentage of 80 or higher is required.

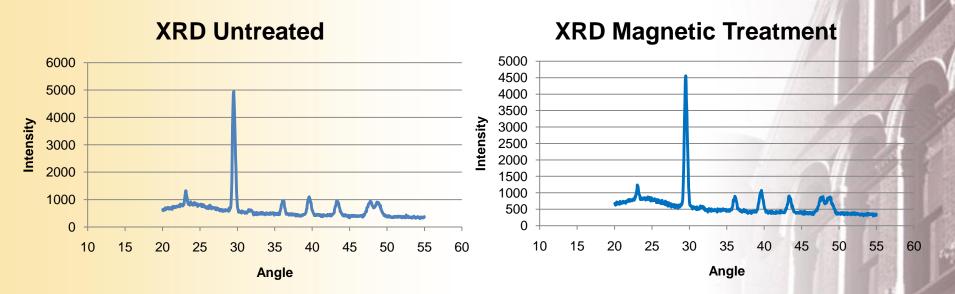


Rapid Test

- Due to the length of time and volume of water needed for the DVGW-W512 protocol, a more rapid testing protocol would be highly desirable.
- Some routes were explored in order to develop a more rapid testing protocol for the scale inducing technologies.



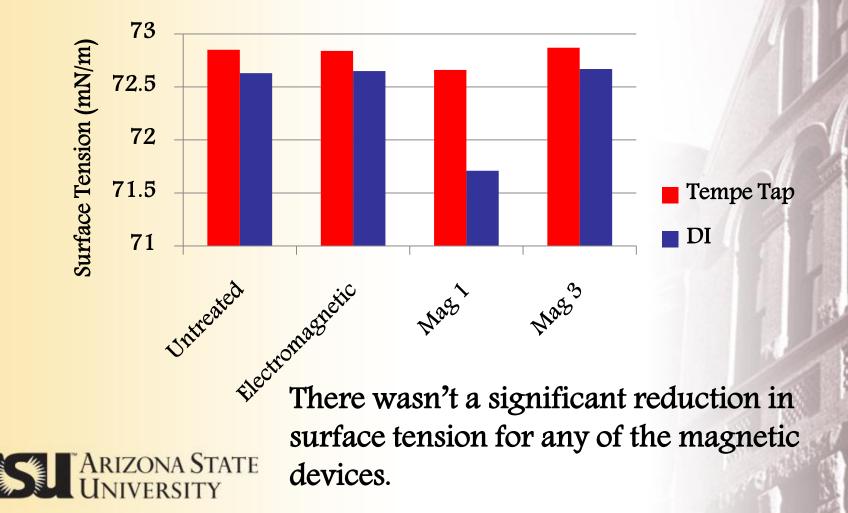
X-Ray Diffraction



Both untreated and magnetically treated cases have calcite patterned peaks

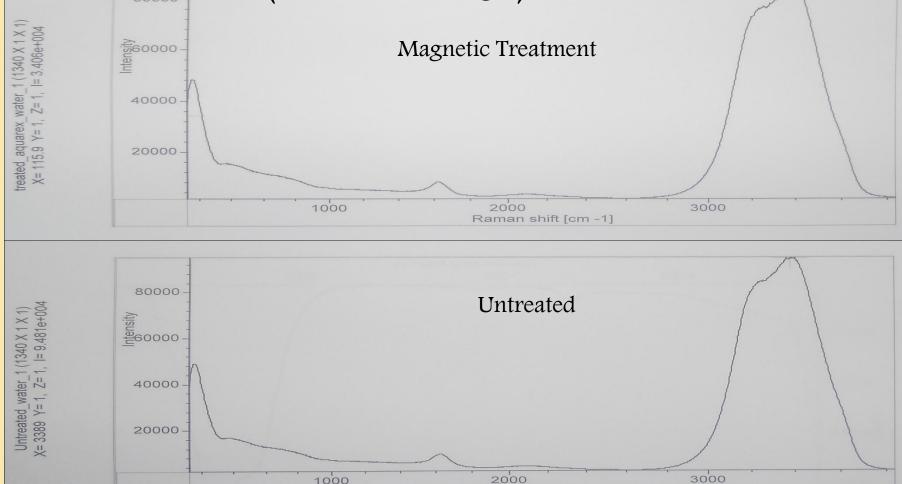


Surface Tension

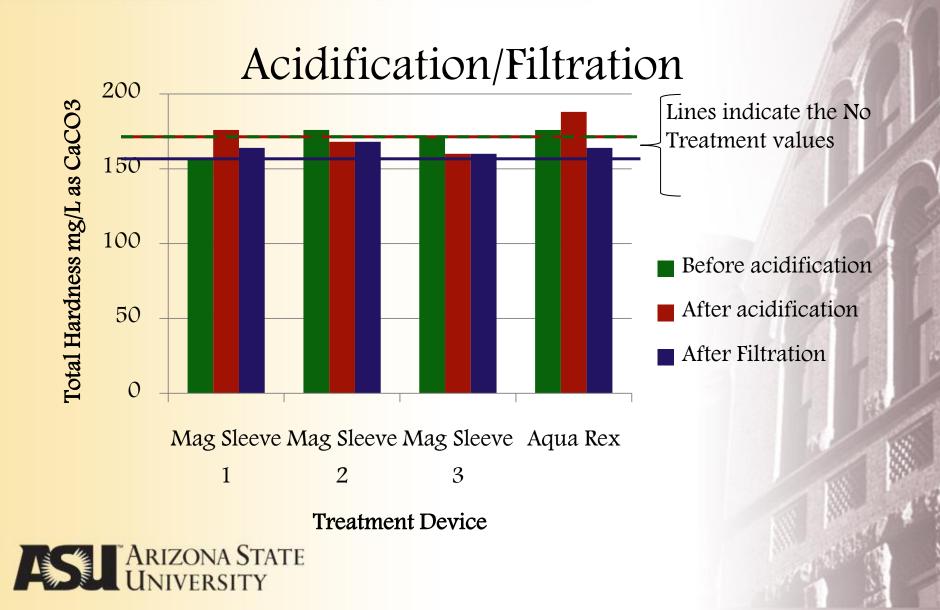




532nm Laser (excitation wavelength), Max Power = 70mW



Raman shift [cm -1]



Conclusions

- All alternative devices were effective at reducing scale.
- The most promising technology is the template assisted crystallization with scale reductions of over 90%.
- Further study is needed to look into the mechanisms at work for the magnetic treatment and a rapid testing protocol.



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Future Work

- Complete testing of CAP canal and Scottsdale groundwater
- Continue exploring possibilities for a more rapid testing protocol
- Consider other no-salt water conditioning devices
- Develop guidelines for consumers such as a rating system to compare water conditioning devices



Aknowledgements

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